3 Review Article

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PROTEINS: THE INGENIOUS WORKHORSE MOLECULES, ADETAILED OVERVIEW.

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9 Abstract

10 Proteins are the most common macromolecules in humans. Proteins are formed from

amino acid units following the template of the genetic code in the protein coding regions

- 12 of the deoxyribonucleic acid (DNA). It has protean structure and function throughout the
- 13 human body.

The amino acid sequence of a protein constitutes its fundamentalor(primary) structure. 14 Proteins' secondary, tertiary, and quaternary structures are determined by their basic 15 16 structure, which represents non-covalent protein folds containing domains and motifs. 17 Protein post-translational modification adds chemical groups to one or more amino acid units, diversifying protein structure and activities. These dictate protein characteristics 18 19 and activities. Protein structure is studied via x-ray crystallography and magnetic 20 resonance imaging. There are various bioinformatics websites and online databases 21 that focus on proteins, protein domains, motifs, and folds, as well as their identification, 22 annotation, classification, and structural alignment. This paper presents amino acid 23 composition, structure, classification, physicochemical properties, domains and folds 24 and motifs of proteins. It also explains the synthesis, post translational modification and 25 other related characteristics of proteins. Research into proteins has aided the 26 understanding of the etiology, diagnosis and treatment of diseases in humans such as 27 cancer, neurodegenerative and metabolic diseases. It has also led to the discovery of 28 drug targets and disease biomarkers. This overview aims to increase these readers' 29 understandingof proteins by deconstructing some of the most common protein-related subjects and titles. It has provided both classical and contemporary knowledge on the 30 31 subject. This shall be of immense benefits to early career researchers and both undergraduate and post graduate students of molecular biology, biochemistry other life 32 33 sciences and even non-life science disciplines.

Keywords Proteins, amino acids, protein structure, protein synthesis, post translational
 modification of proteins.

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39 **1.INTRODUCTION**

40 Proteins are found throughout the human body. They are the most common macromolecules in humans, accounting for more than half of the dry weight of cells 41 (Cozzone, 2002). They are involved in almost every aspect of cell function. They are 42 43 necessary for survival (W.Rose, 2019). In contrast to carbohydrates (stored as glycogen) and lipids (stored as lipid droplets in adipose tissue), proteins lack specific 44 storage depots in the human body (A.Rose, 2019). This is why the body conserves 45 proteins during times of food deficit, such as fasting (Cahill, 2006), by delaying or 46 stopping tissue amino acid use (Carroll, 2015; A.Rose and Richter, 2009; Millward and 47 Garlick, 1972). However, the amino groups of specific amino acids are still used in 48 49 various metabolic reactions and must be replenished by dietary intake (Reed and Biolo, 2002). Since 1912, amino acids have been divided into nutritionally essential 50 (indispensable) and nutritionally non-essential (dispensable) categories based on 51 52 growth and nitrogen balance (Hou et al, 2015).

Nutritionally essential amino acids must be obtained through diet since their carbon 53 skeletons cannot be produced de novo in human cells or are insufficiently generated to 54 55 meet human metabolic needs. (Lopez and Mohuiddin, 2022). The amino acids are phenylalanine, valine, tryptophan, threonine, methionine, histidine, isoleucine, leucine, 56 and lysine. A complete protein includes all of the essential amino acids. Non-essential 57 (dispensable) amino acids can be produced de novo in human and animal cells and 58 59 may not be present in the diet. They are alanine, asparagine, aspartic acid, glutamic acid, and serine. 60

However, Hou et al. (2015) believe that non-essential amino acids must be included in the diet for optimal growth, development, and metabolic outcomes. Furthermore, there is a subset of amino acids known as conditionally essential amino acids because they can be synthesized by human or animal cells in healthy physiological states but not in unhealthy physiological states such as starvation or in-born errors of metabolism, where they must be obtained through diet. They comprise arginine, cysteine, glutamine, glycine, proline, and tyrosine.

The central dogma of molecular biology holds that genes are transcribed into 68 69 messenger RNA, which is then translated into proteins that mediate metabolic networks 70 in creatures including humans. Metabolic networks regulate various aspects of life, 71 including cellular electron transfer in oxidative phosphorylation, muscle 72 tissue contraction, and hearing at the organ level (Salem, 2013). Proteins are complex organic macromolecules made up of amino acid units. Animal protein is made up of 20 73 74 naturally occurring amino acid units.

The amino acid units or residues are connected together by condensation between the amino group of one amino acid and the carboxyl group of the next amino acid unit,

resulting in the loss of a water molecule to form peptide bonds that connect the amino

78 acid units in a linear pattern. Oligopeptides have fewer connected amino acid units than polypeptides. Proteins are polypeptides with molecular weights more than 5000 Daltons 79 (Sumner 1988). Jons Jacob Berzelius, a Swedish scientist, created the term protein in 80 81 1838 from the Greek 'proteios', which means 'holding first place' (Katz, 2021). Proteins differ in amino acid quantities and sequences, making them species and organ specific. 82 The proteome denotes all the proteins that are produced by an organism or a person 83 84 while the genome denotes all the genes that exist in the organism or person. Proteomics systematically studies these proteins which are not static but vary with time 85 Pharmacoproteomics is an exploitation of the great potentials of proteomics even 86 beyond pharmacogenomics in precision medicine to 87 develop and desian pharmacotherapeutics that target a specific phenotype. Single nucleotide polymorphism 88 89 may yield varying mRNAs that in turn yield varying amino acids and proteins as the 90 phenotype (Jain, 2016).

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93 2. AMINO ACID COMPOSITION OF PROTEINS



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97 Amino acids are the building components of proteins. Each amino acid has an amino group, a carboxyl group, and a side chain that extends from the backbone of the 98 polypeptide chain. The attached side chain determines the protein's characteristics. 99 They have different sizes, shapes, polarities, charges, and hydrophobicities. Glycine, 100 alanine, valine, leucine, isoleucine, and methionine all have neutral aliphatic side 101 chains, while phenylalanine, tyrosine, and tryptophan have aromatic side chains. At 102 neutral pH, basic side chains include lysine, arginine, and histidine; acidic side chains 103 include glutamic acid and aspartic acid; and polar uncharged side chains include serine, 104 threonine, glutamine, proline, and cysteine. 105

106 Cysteine has sulfur-containing side chains, which allow disulphide bonds to be formed between polypeptide chain segments or chains. Basic proteins include more basic 107 amino acid side chains than acidic proteins. Proteins, on the other hand, are typically 108 amphoteric, which means they can function as acids or bases due to the presence of 109 110 both acidic and basic amino acid side chains. Protein ionization is pH-dependent. The amino group is positively charged at low pH, whereas the carboxyl group is unionized. 111 112 At high pH, the carboxyl group is negatively charged, whereas the amino group is 113 unionized. pH affects protein staining because it influences protein ionization. The 114 isoelectric point is the pH at which positive and negative charges in proteins are balanced(Sumner, 1988). 115

116 Selenocysteine and pyrrolysine are newly discovered naturally occurring amino acids

⁹⁵ **Figure 1:**Amino acid Structure(Sumner 1988)

found in some organisms. Some amino acids are formed by modifying one of the twenty amino acids and are rarely encountered in proteins. These include hydroxylysine and hydroxyproline in collagen, the neurotransmitter gamma aminobutyric acid derived from glutamic acid decarboxylation, the neurotransmitter serotonin formed from tryptophan, gamma carboxyglutamic acid in calcium binding proteins, histamine derived from histidine decarboxylation, and epinephrine formed from tyrosine. (Garret and Grisham, 2010)

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126 **3.PROTEIN STRUCTURE**

127 There exist a traditional "sequence – structure – function" model in protein discuss 128 which implies that the sequence of amino acids in proteins determine their structure 129 which in turn dictates their functions. (He et al, 2009, Wright & Dyson, 1999). This is 130 captured in the central dogma of protein structural biology which has it that the amino acid sequence of a protein has all the information required for a protein to fold into a 131 132 three-dimensional structure under favorable physiological or experimental conditions and that this structure plays a significant role in specifying the function of the protein (Li 133 et al, 2014). The high-resolution X-ray crystallography is used to study the structure of 134 proteins as it produces static protein structures (Henzler-Wildman & Kern, 2007). 135 Nuclear magnetic resonance (NMR) is also used for this (Sun et al, 2004). However, 136 proteins are dynamic molecules and their physical motions within cells determine their 137 biological functions (Yang, 2014). These physical motions include the protein folding 138 139 process itself that gave rise to the original folded state, the protein dynamics and molecular motions that occurs in the folded conformation and the dynamics of protein 140 ligand interaction(Li et al, 2014). Protein dynamics may be studied by x-ray 141 crystallography, nuclear magnetic resonance (NMR) spectroscopy, Laue x-ray 142 diffraction, cryo-electron microscopy, small angle x-ray scattering, circular dichroism 143 (CD), fluorescence resonance energy transfer (FRET), infrared spectroscopy, Raman 144 spectroscopy, electron paramagnetic resonance and computational approaches like 145 146 molecular dynamics (MD) simulation and their derivatives like simulated annealing, 147 replica exchange Monte Carlo technique, force field simplification models eg Gaussian network models, GO models and enhanced sampling techniques like metadynamics, 148 149 umbrella sampling, targeted and steered simulations (Li et al, 2014).

150 Proteinstructure is grouped into four levels of increasing complexity: the primary structure, secondary structure, tertiary structure and quaternary structure. The primary 151 structure is the simple linearly linked arrangement of amino acids found in newly 152 synthesized polypeptide chains. In the primary structure, covalent bonds link the 153 carboxyl group of the preceding amino acid with the amino group of the next amino 154 acid. The first amino acid in the polypeptide chain is the N-terminus and the last amino 155 acid with the carboxylic acid group is the C-terminus. The primary structure determines 156 the higher orders of protein structure (Anfinsen, 1973). 157

158 The secondary structure of proteins refers to the pleating and spiraling of the primary protein structure due to hydrogen bonds between parts of the same polypeptide 159 160 juxtaposed to each other (Sumner, 1988). This is the rearrangement of the primary structure of protein caused by regular hydrogen bonds between the backbone C=O 161 group and the NH group in protein sequence. Secondary protein structure is further 162 elucidated as the local spatial conformation of the polypeptide back bone excluding the 163 side chains. The α -helix and β -pleats are secondary protein structures (He, et al, 2009) 164 as well as turns (Sun et al, 2004). The α-helix is a right handled spatial helical 165 conformation of the backbone of the protein structure (shape of a helix) whereby the 166 C=O group of one amino acid (ith residue.) is linked to the N-H group of ith+ 4 amino 167 acid residue in advance of the ith residue by hydrogen bonding. There are normally 3.6 168 amino acid residues per turn of the helix. B-pleated sheets or simply B-sheets form a 169 group of secondary protein structure formed by extended strands of polypeptide chains 170 linked by interchain hydrogen bonding. The β sheets or pleats are composed of two or 171 more extended polypeptide chains called β strands that run alongside each other and 172 173 are organized in parallel or antiparallel pattern. The β -pleated sheets may be parallel 174 where the C terminus or N terminus of one end lie side by side or antiparallel where they lie in opposite directions. A mixed β -sheet is formed from both parallel and 175 antiparallel strands. B-turns or β -bends or reverse turns are twisted polypeptide chains 176 that connect helices and β-sheets. Loops and coils are secondary protein structure that 177 have irregular hydrogen bonding from those in reverse turns, β -sheets and α -helices 178 179 (Sun et al, 2004).

180 Topology refers to the pattern of connectivity between secondary protein structures, for 181 instance, β - α - β topology where a β -strand is connected to an α -helix, then to another β -182 strand. The topology diagram is a two-dimensional diagrammatic representation of the 183 secondary protein structure.



Figure 2: Secondary Protein Structure: Alpha Helix (Alberts et al., 2002)



188 **Figure 3:** Secondary Protein Structure: Beta Sheet (Alberts et al., 2002)

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The tertiary structure of proteins is the overall three-dimensional appearance of proteins due to the interactions of the side chains and the way in which the secondary structure packs together to fold the protein. It is the three-dimensional arrangement of all the atoms that comprise the protein molecule. The tertiary structure of the protein molecule is maintained by hydrogen bonds, disulphide bonds, Van Der Waal forces and ionic (electrostatic) bonds that hold previously distant points into juxtaposition.

196 The quaternary protein structure is a multi-subunit protein molecule complex composed 197 of many interacting subunits that are folded to form the complex (He et al. 2009, 198 Sumner, 1988). It is the exact spatial arrangement of subunits within a protein (Sun et 199 al, 2004). It is the interaction of two or more constituent polypeptide subunits which 200 creates the specific structural disposition of the protein molecule like the hemoglobin 201 molecule that has two alpha and two beta subunit polypeptide chains making up its 202 tetrameric structure. Molecular chaperones, intramolecular hydrogen bonds, electrostatic 203 and Van der Waals forces contribute to proper folding of proteins (Upadhyay 2009).

204 The elucidation of protein structure has benefitted from advances in artificial 205 intelligence. The neural network, alpha fold 2 (AF2) was developed by Jumper and his 206 colleagues (Jumper et al., 2021) to predict the three-dimensional protein structure from 207 its amino acid sequence to near experimental accuracy. Alpha fold 3 (AF3) was later 208 developed by Abramson and his colleagues to predict the structure of proteins bound to 209 other complexes like ions, ligands, nucleic acids (Abramson et al., 2024). Protein 210 structure is essential in structure-based drug development and targeted mutagenesis (Tunyasuvunakool et al., 2021) Proteomics is the systematic study and analysis of 211 212 proteins produced by cells of an organism or a human in disease and in the healthy 213 state. This has been applied in the evaluation, monitoring, treatment and prognostication in disease management such is poly trauma (Alpantaki et al., 2007). 214 Proteomics has been applied in precision medicine and the identification of disease 215 biomarkers like thyroglobulin, a biomarker for follicular thyroid carcinoma (Van Eyk and 216 217 Snyder, 2019).

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219 4. PROTEIN DOMAINS, FOLDS AND MOTIFS

Domains, folds, and motifs are all components of the tertiary protein structure. Domains are discrete, compactly folded tertiary protein units that range in length from 100 to 200 residues. According to an examination of protein structures from the protein data bank, autonomously folding protein domains must have a maximum size of 200 residues. Folds are domain components characterized by the spatial organization of protein secondary structural elements into domain-like structures that comprise the domain. Motifs are similar to folds but smaller, and they are subunits of protein folds. Structural and sequence motifs are interchangeable. Sequence motifs are well preserved
 structural motifs with a distinctive signature amino acid sequence.

229 Protein structural motifs include alpha-helix bundles, coiled coils, beta hairpins, Greek 230 key and jelly-roll motifs, beta-sandwich motifs, mixed alpha/beta-sandwich motifs, betabarrels, alpha/beta barrels, and beta propellers. A specific structural motif, fold, or 231 232 domain may be linked to a particular protein activity (Sun et al, 2004). 233 There are various bioinformatics websites and online databases that focus on proteins, 234 protein domains, motifs, and folds, as well as their identification, annotation, classification, and structural alignment. PROSITE, SMART, Pfam, Interpro, TIGRFAM, 235 236 Blocks, CDD, PRINTS, ProDom, RCSB PDB, MSD, MMBD, SCOP, CATH, FSSP, HOMSTRAD, VAST, and CE are among of them. 237

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241 5. THE PHYSICOCHEMICAL PROPERTIES OF PROTEINS

The physicochemical properties of proteins depend on their amino acid constituents. Amino acids except glycine, exist as isomers (mirror images of themselves). The isomers are classed as 'D' (dextrorotatory) for rotation to the right or 'L' (levorotatory), for rotation to the left because they possess the property of optical rotation, which is their ability to rotate the plane of polarized light.Except for a few 'D' amino acid isomers found in bacteria, proteins are virtually entirely composed of 'L' amino acids.

Proteins that contain the aromatic amino acids, phenylalanine, tryptophan and tyrosine 248 249 are intrinsically fluorescent (Lakowicz, 2006). Here we mean that they possess the ability to absorb ultraviolet electromagnetic radiation at certain wavelengths and then 250 251 emit light. Protein fluorescence is applied in the spectroscopic determination of protein Fluorescent proteins are useful for live-cell imaging, whole animal 252 concentration. 253 imaging as well as serving as genetic labels when they are included in transgenic approaches (Kremers et al, 2011). Tryptophan also exhibits phosphorescence meaning 254 that they emit the absorbed light for a longer duration (Garrett and Grisham, 2010). 255 256 Fluorescent proteins have a unique β-barrel fold made up of 11 β-sheets surrounding an 257 α -helix (Ormo et al., 1996). Mutations of fluorescent proteins have the potential of changing their fluorescent properties because of their unique β -barrel fold, making them 258 capable of emitting light of various colors. This increases their utility as molecular 259 probes(Kremers et al, 2011). Fluorescent proteins differ in their brightness, 260 photostability, PH stability and monomeric properties (Cranfill et al., 2016). 261

Solubility is a physicochemical property of a solid substance, liquid, or gas known as a solute that dissolves in a liquid solvent to generate a homogeneous solution (Wedler & Freund, 2012). Protein solubility is the concentration of protein in a saturated solution that is in equilibrium with a solid phase, which can be crystalline or amorphous depending on the conditions (Cohn et al., 1943; Arakawa & Timasheff, 1985). Extrinsic and intrinsic variables regulate protein solubility (Kramer et al., 2012). Extrinsic parameters influencing protein solubility include pH, temperature, ionic strength, and the
 presence of solvent additives (Colowick et al., 1962). Protein solubility is determined
 mostly by the amino acid concentration on its surface (Kramer et al., 2012).

271 The extrinsic factors affecting protein solubility are easier to manipulate than the 272 intrinsic t54t54factors. There is inadequate knowledge of how to alter the intrinsic factors to increase protein solubility (Arakawa & Timasheff, 1985, Volkin & Middaugh, 273 1992).Protein solubility in aqueous solution can be quantified by adding lyophilized 274 275 protein solvent and by the concentration of protein by ultrafiltration(Kramer et al., 2012). Protein solubility may be measured by using precipitants which are extraneous agents 276 that reduce the solubility of proteins. Protein precipitants are classified into three types: 277 salts, organic solvents, and long chain polymers (McPherson 2004). Denatured proteins 278 279 have significantly higher hydrophobicity and lower solubility than native (properly folded) proteins. (Garidel 2013). Proteins are less soluble in polar solvents like ethanol and 280 281 insoluble in nonpolar solvents like cyclohexane (Nick Pace et al., 2004).

Hydration is critical for protein solubility in water. Protein hydration is the process of 282 gradually adding water to dry protein until a hydration point is achieved, after which 283 further water addition makes no change but dilutes the protein (Rupley & Careri, 1991). 284 Water of hydration refers to the water that is attached to the protein. The hydration shell 285 is the water that remains with the protein at the end of hydration (Rupley & Careri, 286 1991). The addition of a salt, such as ammonium sulfate, can diminish the solubility of 287 protein dissolved in water, causing the protein to precipitate. According to Haurowitz et 288 289 al. (2023), the protein is salted off or precipitated.

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291 Proteins are usually identified based on their molecular weight. Atomic force microscopy 292 generates three-dimensional pictures of single molecules, allowing for molecular volume measurements. It can be used to calculate protein molecular weights because the 293 294 molecular volume of single and multimeric proteins correlates with their molecular 295 weights (Schneider et al., 1998). However, ultracentrifugation is the most commonly used method for measuring the molecular weight of proteins. SDS-PAGE is a method 296 297 for determining protein molecular weight by comparing it to the electrophoretic migration of a protein with a known molecular weight (Jardine, 1990). Differences between the 298 measured protein molecular weight and those predicted by its amino acid sequence can 299 300 be attributable to post-translational changes of the protein. (Ahmad et al., 2005).

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303 6. CLASSIFICATION OF PROTEINS

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³⁰⁵ Proteins are conventionally classified according to structure into globular proteins. ³⁰⁶ Fibrous proteins and membrane proteins (Chan& Dill, 1993). Globular proteins are ³⁰⁷ sphere-like and are soluble in aqueous solutions. The hydrophobic monomers are ³⁰⁸ tucked into the core of the protein while the polar monomers reside on the surface. ³⁰⁹ Enzymes that catalyze nearly all biochemical reactions in living cells are globular ³¹⁰ proteins. Globular proteins have β -sheets and α -helices as their secondary ³¹¹ structure(Chan & Dill, 1993). Fibrous proteins are long polypeptide chains that are 312 parallel to one another and are stabilized by cross linkages. They form support 313 structures in humans and are important in biomechanics. They are insoluble in aqueous 314 solutions. Collagen, a fibrous protein, is the most prevalent protein in the human body, 315 accounting for up to 30% of the total protein pool. It possesses a triple-helix structure 316 (Delgoda & Murray, 2017). Membrane proteins are responsible for transport and 317 signaling across cell membranes. They can be transmembrane α -helices or β -strands 318 (Gromiha & Ou, 2014).

319 Proteins may further be classified basedon their function or the biological process they partake in thus: nucleic acid binding protein, transcription factor, receptor, transporter, 320 321 defense/immunity protein, isomerase, lyase, phosphatase, ligase, kinase, protease, 322 oxidoreductase, enzyme modulator, transferase, cell adhesion molecule, extracellular matrix protein, membrane traffic protein, hydrolase, trans-membrane receptor 323 324 regulatory/adaptive protein, chaperone, calcium binding protein, transfer/carrier protein, storage protein, cytoskeleton, viral protein, surfactant and structural protein (Gerret & 325 Grisham, 2010). 326

The modern era of molecular biology and bioinformatics has led to classification of 327 proteins according to structural and functional differences and similarities by adopting 328 curated, automated or a hybrid approach (Ouzounis et al., 2003). The resources or 329 databases for structural classification of proteins include: SCOP (structural classification 330 of proteins); CATH (class, architecture, topology, homology); FSSP (fold classification 331 based on structure-structure alignment of proteins); DSSP (database of secondary 332 333 structure of proteins); HSSP (homology and secondary structure of proteins); Pfam (protein families); PRINTS (protein fingerprints); SMART (simple modular architecture 334 335 research tool); PROSITE; TIGRFAM; PRODOM (PROTEIN DOMAIN); BLOCKS; 336 eMOTIF; SYSTERS; ClustR (clusters of related proteins); COGS (clusters of orthologous groups); ProtoMap; MetaFam (meta database of protein families); TRIBES 337 (Ouzounis et al., 2003). The resources or databases for functional classification of 338 proteins include: EC (enzyme classification); YPD (yeast proteome database); SGD 339 (saccharomyces genome database); MIPS (Munich information center for protein 340 sequences); WIT (what is there) renamed ERGO; STRING (search tools for the retrieval 341 342 of interacting genes/proteins); AllFuse; Predictome; Riley; GeneQuiz; GO (gene ontology); BioCyc; KEGG (Kyoto encyclopedia of genes and genomes); DIP (Database 343 344 of Interacting Proteins); YPL.db; TRIPLE (Transposon-Insertion phenotypes, localization and expression in saccharomyces); MINT (Molecular Interaction); BIND (Biomolecular 345 Interactions Database); PIM (Protein Interaction Manager); CellZome db (Ouzounis et 346 347 al., 2003).

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349 7. PROTEIN IDENTIFICATION BY HISTOCHEMISTRY

Histochemical identification of proteins in animal tissues can be achieved by ultraviolet (UV) absorption; direct staining methods; enzyme histochemical methods if the protein is an enzyme; immunohistochemical methods if the protein is antigenic and
 autoradiography after injecting a radioactively labelled metabolic precursor (Sumner,
 1988).

Proteins absorb UV radiation at peak wavelengths of 240 to 310 nm based on their amino acid content. This coincides with the peak UV absorption of nucleic acids making identification difficult. However, hemoglobin has a peak UV absorption wavelength at 416 nm. It can be quantitatively identified by microspectrophotometry (James & Tas, 1984). UV absorption at wavelength of 280nm is due to tryptophan and tyrosine in the protein sequence and their absence in the protein sequence gives erroneous results (Noble & Bailey, 2009).

362 The direct staining methods for identifying proteins include: methods for staining proteins ionically according to their acidic or basic properties, like the alkaline fast green 363 364 method as an instance; methods for staining chemical groups on a protein by a covalent attachment, like the ninhydrin-schiff test for amino groups (Yasuma & Ichikawa, 1953); 365 methods for fluorescent staining of proteins by covalent bonding to a fluorescent dye, 366 like the fluorescein isothiocyanate (FITC) as an instance (Chaganti et al., 2018); 367 Methods for staining specific amino acid residues like tryptophan using the DMAB-nitrite 368 369 (dehydrocoupling of dimethylamine-borane-nitrite) method (Shori et al., 1997); Methods for staining specific proteins like collagen using picro-sirius red (Puchtler et al., 1988); 370 methods for detecting abolition of staining following removal of proteins by specific 371 372 enzyme digestion (Pearse, 1968).

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374 8. PROTEIN SYNTHESIS

375 Proteins synthesis in cells uses amino acid building blocks and messenger ribonucleic 376 acids (mRNA) as the templates. Ribonucleic acid (RNA) is important in converting the genetic information into proteins. It differs from DNA (deoxyribonucleic acid): It has only 377 one strand and is not double stranded like DNA; It has a ribose sugar which differs from 378 379 the deoxyribose sugar of DNA in having an extra hydroxyl group; It has an uracil nitrogenous base instead of the thymine of DNA, though the uracil still binds with 380 adenine as its complementary base. Three types of RNA are involved in the conversion 381 of genetic information to proteins: messenger RNA (mRNA), transfer RNA (tRNA) and 382 383 ribosomal RNA (rRNA) (Li et al., 2011).

384 Gene expression proceeds through two stages: transcription and translation. 385 Transcription is the process of converting genetic information from DNA sequence to 386 mRNA. Messenger RNA carries the genetic information from the chromosome to the 387 site of protein synthesis.

388 Translation is the process of converting the genetic information in mRNA to proteins by

polypeptide synthesis(Kohler et al., 2017; Artsimovitch, 2018).

Transcription is the process of copying a gene's sequence from DNA into a singlestranded mRNA molecule. Transcription has three stages: initiation, elongation, and termination.

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Initiation: The RNA polymerase enzyme binds to the DNA segment that will be transcribed in a location known as the promoter and opens the double helix. The promoter region is located in front of the gene that will be transcribed and is typically made up of a string of adenine and thymine bases that act as the recognition region for the RNA polymerase. The promoter specifies which DNA strand should be transcribed and where RNA polymerase should begin transcription on the DNA. RNA polymerase's binding site recognizes only the promoter region and can only bind in front of the gene.

Elongation: The RNA polymerase attaches to the DNA promoter region and opens the double helix. It then creates the single-stranded mRNA in a 5' to 3' direction without transcribing the promoter region, analogous to DNA replication. However, RNA polymerase does not need a primer and merely transcribes one strand of DNA. The transcribed DNA strand acts as a template strand. The transcribed mRNA strand is complementary to the template strand of DNA, with the exception that it contains uracil instead of thymine.

Termination: The synthesis of the mRNA goes on until the RNA polymerase reaches the end of the gene which is recognized by a stop sequence called the termination sequence. Here transcription stops, the newly synthesized mRNA disconnects from the DNA template strand. The RNA polymerase is free to bind to another promoter region

- 412 and transcribe another mRNA
- 413 (Merrick, 1992).

In the second part of the central dogma of molecular biology, the genetic information in the mRNA is decoded to make chains of amino acids to form polypeptides. Translation involves protein synthesis and is dependent on the nature of the genetic code. All proteins are made up of only 20 amino acids, which are coded for by various combinations of the four nitrogenous bases. The genes in the DNA are read in groups of three nucleotides.

- 420 Codons can be any combination of the four nitrogenous bases, resulting in a total of 64
 421 distinct codons (43 = 64) for the 20 amino acids. Codons apply to both DNA and mRNA.
 422 In the mRNA codon, AUG codes for the amino acid methionine and is known as
- the start codon' since all translations begin there. So, while all polypeptide chains begin
 with methionine, it may be edited out later. Three more codons (UAA, UAG, and UGA)
 are known as "stop codons," and they do not code for amino acids. They indicate where
- 426 the translation should conclude.
- The remaining sixty codons code for amino acids implying that more than one codon, in fact, many codons may code for the same amino acid eg serine and leucine each has 6 codons.
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- 431 Translation has three stages like transcription: initiation, elongation, termination.
 432 Ribosome recycling follows this (Maguat et al., 2010).

433 Initiation: Initiation of translation starts when a ribosome recognizes a unique mRNA codon and binds to it. The ribosome in eukaryotic cells has a large subunit and a small 434 435 subunit. The two subunits bind to the mRNA, clamping it between them and move along it in a 5' to 3' direction adding new amino acid units to the growing polypeptide chain 436 each time it reads a codon. The ribosome synthesizes different polypeptides by reading 437 the coding sequence on the mRNA. It must begin reading this codon sequence from the 438 start codon AUG, in order to read the right reading frame of the mRNA. The ribosome 439 440 binds to the mRNA's start codon, ensuring that the genetic code is translated using the mRNA's reading frame. To translate the genetic code into the correct amino acid 441 sequence, the mRNA must be positioned in the ribosome in its reading frame. Once the 442 ribosome has attached to the mRNA molecule, transfer RNA (tRNA) provides the 443 appropriate amino acid to be added to the polypeptide. The anticodon, a three-base 444 sequence, is found at one end of the tRNA and is complementary to the mRNA 445 sequence. The opposite end of the tRNA transports the matching amino acid to the 446 mRNA sequence. Each tRNA transports a unique amino acid. There are 20 amino acids 447 448 and 64 codons, hence the number of tRNA molecules available varies from 20 to 64 depending on the organism. The tRNA has a clover leaf structure (Laursen et al. 2005). 449

Elongation: The ribosome recognizes the start codon, AUG, which also codes for 450 methionine. All polypeptides begin with the amino acid methionine. The ribosome has 451 two tRNA attachment sites: the 'A' (aminoacyl) and the 'P' (peptidyl). The tRNA 452 453 containing anticodon corresponding to the mRNA's start codon enters the P site. The next tRNA that has the designated amino acid enters the A site. A peptide link is formed 454 between the methionine and the following amino acid. The ribosome moves over one 455 codon, freeing up the P site for the second tRNA and releasing the methionine-carrying-456 tRNA. The second tRNA goes to the P site, freeing up the A site for the third tRNA. The 457 released tRNAs are recycled in the cytoplasm by incorporating new amino acids. This 458 process continues until the full mRNA sequence is translated into the specified protein 459 460 and the ribosome reaches a stop codon (Czworkowski & Moore, 1996).

Termination: The ribosome eventually hits one of the stop codons UGA, UAG, or UAA, which do not code for any amino acid and thus lack a corresponding tRNA. The release factor is a protein that detects stalled ribosomes and aids in the release of the polypeptide chain. The ribosome's two subunits separate from the mRNA as translation ends.



467 Figure 4: Protein Synthesis (Plotkin & Onuchic, 2002).

468 9. POST-TRANSLATIONAL MODIFICATION OF PROTEINS

469 Post-translational modification (PTM) of proteins are alterations in the structural content of proteins after their biosynthesis by proteolytic cleavage and incorporation of one or 470 471 more modifying groups like phosphoryl, glycosyl, acetyl, methyl, etc to one or more 472 amino acids, consequentially changing the properties and functions of the protein 473 (Ramazi et al., 2020). Proteins can be reversibly or irreversibly modified after they have 474 been translated (Wang I et al., 2014). A single amino acid or a group of amino acids 475 may be modified (Huang et al., 2019). Post-translational modification of proteins may 476 occur in the nucleus, cytoplasm, endoplasmic reticulum or golgi apparatus (Blom et al., 477 2004). There are over 400 PTMs of proteins (Khoury et al., 2011). Phosphorylation, acetylation and ubiguitination comprise more than 90% of PTMs reported in the dbPTM 478 479 , a comprehensive database of PTM of proteins. Each amino acid has at least three different PTMs. Lysine has fifteen PTMs, the highest number of PTMs among amino 480 Other major PTMs include glycosylation, methylation, 481 acids. SUMOvlation. palmitoylation, myristoylation, prenylation, sulfation (Ramazi & Zahiri, 2021). New PTMs 482

483 discovered by updated mass spectroscopy include ufmylation, crotonylation, succinylation and lactylation (Wu et al., 2024). Post-translational modification of proteins 484 485 may influence biological processes like signal transduction, cell cycle control, gene expression control, gene activation, DNA repair. (Wang II et al., 2015, Strumillo & 486 Beltrao, 2015, Wei et al., 2017), The influence of PTMs of proteins on biological 487 processes is brought about by their effects on protein folding (Del Monte & Agnetti, 488 2014), protein lifespan and protein-protein interaction (Marshall, 1993), protein 489 490 localization(Audagnotto & Dal Peraro, 2017), enzyme function and assembly (Ryšlavá et al., 2013), protein solubility (Ramazi & Zahiri, 2021), molecular trafficking (Cundy et 491 492 al., 2002, Ohtsubo & Marth, 2006), cell-cell and cell-matrix interaction(Goulabchand et al., 2014, Caragea et al., 2007), receptor activation (Ohtsubo & Marth, 2006).Post-493 translational modification of proteins may be studied or detected by proximity ligation 494 495 assay, an immunoassay technology employed in the study of PTM of proteins (Leuchowius et al., 2011), immunoprecipitation which is used in PTM detection assays 496 497 (Fuchs & Strahl, 2011) or a combination of mass spectrometry and immunoprecipitation 498 (Larsen et al., 2006). Other experimental methods for detection of PTMs include Liquid 499 chromatography (Welsch & Nelsestuen, 1988), radioactive chemical method (Slade et al., 2014), western blotting (Jaffrey et al., 2001) and eastern blotting (Welsch 500 & 501 Nelsestuen, 1988). Nanopore technology is emerging as a novel method of detecting PTMs (Zhao et al., 2024). Computational methods are being used in recent times to 502 predict PTMs of proteins because experimental identification of PTMs is 503 expensive(Hasan & Khatun, 2018, Sobolev et al., 2014). Aberrant PTM pathways in 504 505 humans are associated with diseases like cancer, heart failure, autoimmune diseases, neurodegenerative disease like Parkinson's disease, Alzheimer's disease and 506 507 Huntingtons disease (Ramazi et al., 2020).

508

509 **10. CONCLUSION**

Proteins are the most numerous biomolecules in animals and they are the work horse of 510 their life processes. Proteins are polymers made up of amino acid units whose 511 sequence form their primary structure. The protein secondary, tertiary and guaternary 512 structures are as a result of protein folding by interplay of non-covalent bonding. The 513 primary protein structure dictates the subsequent structural folds, properties and 514 functions of the proteins. Proteins are synthesized in cells using the genetic code. Post-515 translational modification of proteins alters their structure and diversify their functions. 516 Their pathway disruption may cause disease. 517

518

519 **COMPETING INTERESTS**

520 The authors declare that they have no competing interests.

- 521
- 522

523	ABBREVIAT	ΓIONS
524	DNA:	Deoxyribonucleic acid
525	RNA:	Ribonucleic acid
526	NMR:	Nuclear magnetic resonance
527	CD:	circular dichroism
528	FRET:	Fluorescence resonance energy transfer
529	MD:	Molecular dynamics
530	D:	Dextrorotatory
531	L:	Levorotatory
532	mRNA:	Messenger ribonucleic acid
533	pH:	Potential of hydrogen
534	SDS-PAGE:	Sodium dodecyl sulfate polyacrylamide gel electrophoresis
535	SCOP:	Structural classification of proteins
536	CATH:	Class, architecture, topology, homology
537	FSSP:	Fold classification based on structure-structure alignment of proteins
538	DSSP: databa	se of secondary structure of proteins
539	HSSP: Homo	logy and secondary structure of proteins
540	Pfam:	Protein families
541	PRINTS:	Protein fingerprints
542	SMART:	Simple modular architecture research tool
543	PRODOM:	Protein domain
544	ClustR:	Clusters of related proteins
545	COGS:	Clusters of orthologous groups
546	MetaFam:	Meta database of protein families
547	EC:	Enzyme classification
548	YPD:	Yeast proteome database
549	SGD:	Saccharomyces genome database

MIPS: Munich information center for protein sequences

551	WIT:	What is there
552	STRING:	Search tools for the retrieval of interacting genes/proteins
553	GO:	Gene ontology
554	KEGG:	Kyoto encyclopedia of genes and genomes
555	DIP:	Database of Interacting Proteins
556	TRIPLE:	Transposon-Insertion phenotypes, localization and expression in saccharomyces
557	MINT:	Molecular Interaction
558	BIND: Biomo	olecular Interactions Database
559	PIM:	Protein Interaction Manager
560	UV:	Ultraviolet
561	FITC:	Fluorescein isothiocyanate
562	DMAB-nitrit	te:Dehydrocoupling of dimethylamine-borane-nitrite
563	tRNA:	Transfer RNA
564	rRNA: Ribos	omal RNA
565	A:	Aminoacyl
566	P:	Peptidyl
567	PTM:	Post-translational modification
568		
569	Disclaimer (Art	<mark>:ificial intelligence)</mark>
570	Option 1:	
571	Author(s) here	by declare that NO generative AI technologies such as Large Language Models (ChatGPT,
572	COPILOT, etc.)	and text-to-image generators have been used during the writing or editing of this
573	manuscript.	
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575	Author(s) here	by declare that generative AI technologies such as Large Language Models, etc. have been
576	used during th	e writing or editing of manuscripts. This explanation will include the name, version,
577	model, and sou	urce of the generative AI technology and as well as all input prompts provided to the
578	generative AI t	echnology
579	Details of the A	Al usage are given below:

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588	REFE	RENCES
589 590 591 592 593 594 595 596 597 598 597 598 599 600 601 602 603 604	1. 2. 3. 4. 5.	 Abramson, J., Adler, J., Dunger, J., Evans, R., Green, T., Pritzel, A., & Jumper, J. M. (2024). Accurate structure prediction of biomolecular interactions with AlphaFold 3. <i>Nature</i>, 1-3. Ahmad, Q. R., Nguyen, D. H., Wingerd, M. A., Church, G. M., & Steffen, M. A. (2005). Molecular weight assessment of proteins in total proteome profiles using 1D-PAGE and LC/MS/MS. <i>Proteome science</i>, <i>3</i>(1), 6. <u>https://doi.org/10.1186/1477-5956-3-6</u> Alberts, B., Johnson, A., Lewis, J., Raff, M., Roberts, K., & Walter, P. (2002). The shape and structure of proteins. In <i>Molecular Biology of the Cell. 4th edition</i>. Garland Science. Alpantaki, K., Tsiridis, E., Pape, H. C., & Giannoudis, P. V. (2007). Application of clinical proteomics in diagnosis and management of trauma patients. <i>Injury</i>, <i>38</i>(3), 263-271. Anfinsen, C.B. 1973. Principles that govern the folding of protein chains. Science 181:223-230.
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